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s.no. 316 BATCH: 87-2	2016
END OF SEMESTER EXAMINATIONS, NOV	<b>90</b> EMBER - 2018

MAJOR : B.SC (CHEMISTRY)

TIME : 3 HOURS

SEMESTER: IV MAX.MARKS: 75

# $\underline{SECTION} - \underline{A} (10 \times 1 = 10)$

GENERAL CHEMISTRY - IV SUBJECT CODE: 11UACH04

# Answer ALL the questions:

- 1. Why transition elements do shows variable oxidation state?
- 2. How do you explain anomalous electronic configuration of  $Cu(4S^{\dagger}3d^{\dagger 0})$ ?
- 3. Draw the structure and give the IUPAC name for any two dihydric phenols.

4. Give the product of the reaction.

- 5. What is Jones-reagent?
- 6. Name the product of cleavage of pinacol with lead tera-acetate.
- 7. What are the Criteria for phase equilibrium?
- 8. Define chemical potential.
- 9. Express Raoult-s law for non-ideal solution in an equation.
- 10. How is activity related to activity-coefficient?

### $\underline{SECTION - B (5 X 4 = 20)}$

### Answer ALL the questions:

11.a) Write a note on Application of CHEMDRAW. (4)

(OR)

- b) Explain irregular variation of ionization first and second enthalpies transition elements. (4)
- 12.a) What is Riemer-Timamann reaction? Give the probable mechanism of Reimer-Tiemann reaction. (4)

(OR)

- b) Discuss Gattermann reaction and its mechanism. (4)
- 13.a) Explain the order of basicity of primary, secondary and tertiary amines. (4). (OR)
  - b) What is Hofman degradation reaction? Explain with suitable example. (4)
- 14.a) Derive Gibbs phase rule.(4)

(OR)

- b) What is meant by relative lowering in vapour pressure? How does it enable determine molecular weight of non-volatile solute? (4)
- 15.a) Explain with diagram the upper critical solution temperature of phenol-wat system. (4)

(OR)

b) Give a brief account on basic principle of steam distillation.(4)

 $\underline{SECTION - C (5 \times 9 = 45)}$ 

#### Answer ALL the questions:

- 16.a) Illustrate variable oxidation state of transition metals and their features. (9). (OR)
  - b) Give an account on different data types in C programming. (9).
- 17.a) Describe the mechanism of the following reactions.
  - (i) Kolbe's reactions (4)
  - (ii) Houben-Hoesch reaction (5)

- b) Write the structure of the product of the following reactions.
  - (i)  $\alpha$  Napthal treated with alkaline  $KM_{\mu}O_4$  (2)
    - (ii) Reduction of  $\beta$  Napthal by Na/Isopentanol (2)
    - (iii) Oxidation of  $\alpha$  Napthal by  $H_2CrO_4$  (2)
    - (iv) Explain why phenol is more acidic than ethanol? (3)
- 18. a) (i) Write a note on Gabriel synthesis of primary amine. (4)
  - (ii) Explain Hinsenberg method for separating primary, secondary, and tertiary amine. (5)

(OR)

- b) (i) How methylamine is converted into Diazomethane? (2)
  - (ii) Complete the following reactions (2)

- (iii) How can we separate primary secondary and tertiary amines by Hofmann's method? (5)
- 19.a) Explain the application of phase diagram for one component system taking water as example. (9)

(OR)

- b) Derive Clausius Clapeyron equation (9)
- 20.a) Describe Azeotropes clearly with phase diagram of two examples. (9)
  - b) Derive Nernst distribution law. (9)