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S.No. 203

BATCH: 87-2011, 2013, 2014

END OF SEMESTER EXAMINATIONS, APRIL / MAY - 2017
PHYSICAL CHEMISTRY - I
SUBJECT CODE: 13UACH07

MAJOR: B.Sc. CHEMISTRY TIME : 3 HOURS SEMESTER: V MAX.MARKS: 75

SECTION - A (10 X 1 = 10)

Answer ALL Questions;

- 1. What is rate constant?
- 2. Define order of a reaction.
- 3. What is meant by activation energy?
- 4. Write the relationship between free energy of activation and entropy of activation.
- 5. Define quantum yield.
- 6. State Grotthus law of photo Chemistry.
- 7. Write any two differences between Chemisorption and physorption.
- 8. What is point group?
- 9. Give the significance of ψ and ψ^2 .
- 10. What are Eigen values and Eigen functions?

SECTION - B (5 X 4 = 20)

Answer ALL questions:

11. a) What is meant by rate of reaction? How is it expressed? Explain.

(or)

- b) Derive an expression for zero order reaction.
- 12. a) Explain Lindemann theory of Unimolecular reactions.

(or)

- b) Compare collision theory with absolute reaction rate theory.
- 13. a) Write the Comparision between thermal and photochemical reaction.

(or)

- b) Discuss any four applications of photochemistry.
- 14. a) Explain diagrammatically that H_2O molecule is Abelian where as NH_3 molecule is non abelian.

(or)

- b) Give the symmetry operation for NH, molecule.
- 15. a) What is meant by black body and black body radiation?

(or)

b) Explain Compton effect.

SECTION - C (5 X 9 = 45)

Answer ALL questions:

16. a) What is a first order reaction? Derive its expression for rate constant.

(or)

- b) A first order reaction is 40 % complete in 50 minutes. Calculate rate constant. In what time will the reaction be 80 % complete.
- 17. a) Describe the collision theory of Unimolecular reactions.

(or)

- b) Discuss the theory of absolute reaction rate.
- 18. a) Derive the rate expression for the photochemical reaction between H_2 and Br_2 .

(or)

- b) Explain the terms
 - i) Fluorescence
- ii) Phosphorescence.
- 19. a) Explain Freundlich adsorption isotherm.

(or)

- b) Explain the following with example
 - i)Axis of Symmetry
 - ii) Plane of Symmetry.
- 20. a) Derive and Discuss E and ψ of a particle in one dimensional box.

(or)

b) Derive Planck's radiation law.