		Reg. No.
o: <b>\49</b>	BATCH: 2014 Colo	

S.No

END OF SEMESTER EXAMINATIONS, APRIL / MAY - 2019

PHYSICAL CHEMISTRY - 1 SUBJECT CODE: 15P3CH03

MAJOR: M.Sc. (CHEMISTRY)

TIME : 3 HOURS SEMESTER : [

MAX. MARKS :70

# $\underline{SECTION} - A (10 \times 1 = 10)$

# Answer ALL the questions:

- 1. What are the different types of operators?
- 2. What is eigen value?
- 3. What is Variation method?
- 4. What is meant by LCAO?
- 5. What is symmetry operation?
- 6. What is similarity transformation?
- 7. What is Gibb's free energy?
- 8. State Planck's statement.
- 9. Define chemical potential.
- 10. What is fugacity?

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#### SECTION - B (5 X 4 = 20)

## Answer ALL the questions:

11. a) Explain the time dependent schrodinger wave equation.

(OR)

- b) Discuss the failure of classical mechanics.
- Apply HMO treatment to ethylene system.

- b) Explain Born Oppenheimer approximations.
- 13. a) What is reducible representations?

- b) Explain Matrix representation of a point group.
- 14. a) Discuss the significance of thermodynamic probability.

(OR)

- b) Describe entropy change accompanying phase transformation.
- 15. a) Derive Duhem Margules equation.

b) How partial molar property is determined?

### $\underline{SECTION - C (5 X 8 = 40)}$

#### Answer ALL the questions:

16. a) Deduce the solution of schrodinger wave equation for a particle in three dimensional

(OR)

- b) Discuss the hermite polynomials and recursion formula.
- 17. a) Apply Huckel theory to Benzene system.

(OR)

- b) How ground state energy of the atom is determined by perturbation method.
- 18. a) Discuss on the electronic spectra of formaldehyde using group theory.

(OR)

- b) Construct the group multiplication table of C<sub>3</sub>V point group.
- 19. a) Discuss on Nernst heat theorem.

(OR)

- b) Derive Gibb's Helm holtz equation.
- 20. a) Determine the activity and activity co-efficient by emf method.

b) How does fugacity varies with temperature and pressure.